STEREOSELECTIVE TOTAL SYNTHESIS OF NATURAL PHYTOL VIA DOUBLE BOND REDUCTIONS BY **BAKER'S YEAST***

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ABSTBACT.- Natural *(E)-(?B,llg)-phytol (J), a component of the chlorophyll molecule, vi* **tamin Kl (2)** *and other natural compounds, was synthesised in enantiomerically and diastereomerically pure form by Li2CuCl4-induced coupling of two Clo-units (2, 12). Both these units were prepared from geraniol (2) via enzymatic enantioselective hydrogenation of activated double bonds.*

Phytol (2) is universally distributed in green plants as a component of the chlorophyll molecule, in which it is present in ester combination, and as a **component part of phylloquinone (Vitamin K1, 3) and tocopherols (Vitamin E, e.g. +tocopherol). It is the lipophilic phytyl side-chain of chlorophyll & which renders the molecule biologically active'. In addition, the antihemorrhagic activity of vitamin Kl is strictly dependent on the (E)-configuration of the** double bond in the phytyl side-chain². Natural phytol has $(E)-(7R,11R)$ **configuration3 and it has been largely used for the synthesis of tocopherols' and natural vitamin K15. Phytol can only be obtained in relatively small quantities from natural sources, and so, a synthetic substitute would be desirable. Many syntheses of phytol in its racemic form6 as well as in the** optically active one^{3,7} have been published.

'Part 4 of the series "Microbial-mediated syntheses of EPC". For Part 3 see ref.ila.

In this paper we describe a new synthesis of $(\underline{E})-(7\underline{R},11\underline{R})$ -phytol in its enantiomerically and diastereomerically pure form. This method is based on the coupling of two C10-units, shown by the disconnection in formula 1. Both the stereogenic centers at the 11- and 7 -position of 1 were obtained through baker's yeast hydrogenation of the "activated" double bond in geraniol (2) and in the Clo-intermediate 4 (Scheme). A sequence of reactions was devised to assure that the correct configuration (E) of the double bond in the starting geraniol was retained in 1.

 (\underline{R}) -Citronellol (5) , prepared by enzymatic reduction of geraniol⁸, was hydrogenated over platinum oxide to give dihydrocitronellol (6) , which was transformed into the bromide 'J by reaction with N-bromoauccinimide and triphenylphosphine. The bromide 7 , obtained in 85% yield from 5 , was then converted with magnesium turnings in tetrahydrofuran into the Crignard derivative 8, which was used for the coupling reaction with the other C_{10} -unit (12) .

To obtain 12, geraniol (3) was oxidized to methyl geraniate (\underline{via} geranial) by the Corey's procedure⁹, which gives a high yield without cia-trans isomerisation of the α , β -olefinic linkage. Methyl geraniate was functionalized by selenium dioxide oxidation using the Rapoport method¹⁰, to produce $\frac{4}{3}$. The double bond in α , β -position with respect to the aldehyde group was then hydrogenated \mathbf{via} baker's yeast fermentation to afford the corresponding α -methyl saturated alcohol (9)¹¹ in enantiomerically pure form and (\S) -configuration^{11a} (35% yield). The double bond in α, β -position with respect to the ester group was retained in its original (E) -configuration during this reduction. In order to have the C10-chiron properly functionalized for the subsequent coupling reaction, the alcoholic function of 9 was converted into the tosylate group, which is the best leaving group for the coupling¹². The carbomethoxy group was reduced with lithium aluminum hydride and the reeulting allylic alcohol **was** protected by reaction with ethyl vinyl ether in trifluoroacetic acid (54% yield of 12 from 9). Coupling of 8 and 12 to produce 13 was then performed with dilithium tetrachlorocuprate at -30°C. Pure $(7R, 11R)$ -phytol (1) was finally obtained by the deprotection of the alcoholic function of 13 by means of HCl 0.5N. No isomerisation of the double bond was found to occur when ueing this acid treatment, **aa** demonstrated by 1H- and 13C-NHR data^{7e.13a.b}. Concerning the C-7 and C-11 stereogenic centers, the diastereomeric purity of 1 was shown to be $>90\%$ by high field ¹³C-NMR analysis^{13c}. This diastereomeric excess of the target compound 1 is also indicative of its enantiomeric excess¹⁴ (>99%), both being directly related to the e.e. of 5 (>98%) and 9 (>97%).

SCHEME

EXPERIMENTAL

IR spectra were recorded on a Perkin-Elmer 681 spectrophotometer. ¹H-NMR spectra were obtained on a Bruker WP80 SY and ¹³C-NMR spectra on a Bruker CXP300 operating at 75.47 MHz. Chemical shifts are reported in 6 from internal MetSi. Optical rotations were measured in a 1.0 dm cell on a Perkin-Elmer Model 241 polarimeter. MS spectra were recorded on a Varian MAT 112 mass spectrometer. Flash column chromatography was performed on silica gel Merck 60 (230-400 mesh). Baker's yeast was "Distillerie Italiane" brand from Eridania [S.Quirico-Trecasali (Parma)] . "Usual work up" means that the reaction mixture was treated with water and CHCls or ether, the organic layer washed with water and brine, dried (HgSOa), and evaporated under vacuum. Phytol (mixture of isomers, Fluka) was used as a reference product.

<u>Methyl 3.7-Dimethyl-8-formyl-2.6-octadienoate</u> (<u>4</u>). *Methyl geraniate (prepared by oxidation of geranlol y& geranlal according to E.Corey's procedure)8(l.44g, 7.9mmol) in EtOH (20mL) was treated dropwlse with a solution of SeOz (1.34g, 12mmol) in EtOH (30rL) at 50-C and refluxed for 5h. The reaction mixture was filtered and the solvent evaporated. Usual work up gave crude 3, which was flash chromatographed (hexane:ethyl acetate 2:l) to afford pure a (760mg, 48% yield): IR and NMR data as in ref.lla; MS* m/e *(relative* **intensity):** *196 I/4+. 1). 167 (10). 163 (84), 137 (25), 136 (2b), 133 (521, 124 (26), 108-i32), 106 i78j; 82 (is);-81 (loo), 40 (75). Anal. Calcd for CllHl* SOS: C, 67.32; *HI 8.22. Found: C, 67.54: H. 8.19.*

 $Method (E) - (S) - (-)-3.7-Di**achyl-8-hydroxy-2-octenoate** (9). *Å was reduced by*$ </u> *baker's yeast fermentation as in ref. lla (35% yield, e.e.>97%): [«]²⁵p
-10.40° (c 2.092, CHCl3); IR and NMR data as in ref. lla; MS m/e (relative intensity): 200 (M*, l), 182 (3). 169 (9). 123 (17). 110 (36). 82 (64). 67 (43). 59 (40), 55 (84), 4i (160). Anal; Calcd for C11HzoOj: C, 65.97; H, 10.66. Found: C, 66.18; H, 9.86.*

 $Methods (E) - (S) - (+) - 3.7 - Dimethyl-8 - [(p-tolylsulfonyl)oxyl-2-octenoate (10).$ </u> *p-TsCl* (700mg, 3.93mmol) *was added to a stirred and ice-cooled solution gf2 (588mg, 2.94mmol) in dry pyridlne (1OmL). The mixture was stirred overnight* at O'C *under nitrogen. Subsequent work up gave* a *crude product which was purified by flash chromatography Ibenrene: ethyl acetate 4:l) to afford 10 (837mg, 80% yield): [ulz5e +0.90' fc 1.69. CHCl3): IR Iliauid film): 2920, 1700,~1635. 1585. 1420, 1345, 1210, 117j, 1160;.1140 cm"; 'H-NMR (CDCl;) 8 b.9 (3-R, d;* $J=6.5$ Hz, C_{H} ₃CH), 1.0-2.2 (7 H, m, 3CH₂ and CH), 2.1 (3 H, d, <u>J</u>=1.5 Hz, *CHsC=), 2.4 (3 H, s, CJL3-Ar'), 3.7 (3 H, s, OCHs), 3.85 (2 H, d, J=5.5 Hs, CH*₂OTs), 5.6 (1 H, q, \bar{y} =1.5 Hz, *CH=*), 7.3 (2 H, d, <u>J</u>=8 Hz, 2Ar <u>ortho</u> to CH3),
7.8 (2 H, d, <u>J</u>=8 Hz, 2Ar <u>ortho</u> to SO3R); MS m/e (relative intensity): 354 (M*, *1), 327 (31), 307 (61, 172 (6), 165 (loo), 160 (661, 150 (72), 139 (24), 135 (4% 127 (72), 124 (93), 91 (97).* Anal. Calcd *for CiaHzsSOs: C, 60.99; Ht 7.39. Found: C, 61.20; H, 7.41.*

rl)ox~l **-** *2* **_** *octen* **1 01 _** *ted portionwise with i~AkidL~104mg. 2.82m&l) and stirred under* **nitrogen at -5'/O'C for** *2h. Subsequent work up ga;e oil, which was purified by flash chromatography (bensene:ethyl* acetate 4:l) to *afford lJ (353mg, 77% yield): [alzs~ t2.6' (c 1.78, CHCl3); IR (liquid film): 3620-3160, 2940, 1360, 1190, 1180, 1100 cm-l; 'H-NMR (CDCl3) 8 0.9 (3 H, d, J=S.5 Hs, CH3CH), 1.6 (3 H, s, CH3C=), 1.0-2.2 (8 H, m, 3CH2, CH and OH), 2.4 (3 H, s, CH3Ar), 3.85 (2 H, 2d, <u>J</u>=5.5 Hz, C<u>H</u>2OTs), 4.15 (2 H, d, J=7 He, CHZOH), 5.35 (1 H, broad t, J=7 Hz, =CflCHzOH), 7.3 (2 H, d, J=8 Hz,* **an** *2Ar Q&& to CH3), 7.8 (2 H, d, ~=8 Hz, 2Ar ortho to SOsR); MS m/e (relative intensity): 326 (M+, l), 308 (3), 241 (Z), 171 i88), 170 (24), 155 (15), 154 (55), 153 (39), 138 (44), 135 (381, 91 (58), 70 (loo), 56 (72). Anal. Calcd for Cl 7HzsSO4: c, 62.55; H, 8.03. Found: C, 62.33; H, 8.10. IE)-(S)-(+)-8-(1'-Ethoxy)ethoxy-2.6-Dimethyl-6-octenyl p-Toluene-sulfonat*
(12). *11 (230mg, 0.7mmol) was treated with ethyl vinyl ether (6.5mL) and CF3COOH (2ml) and stirred at room temperature for 30min. After addition of triethylamine (O.lmL) the* **reaction mixture was** *stirred for 45'. Subsequent work UP gave* a *crude Droduct. which was Durlfied by flash chromatoflraDhy /benzene:ethvl acetate 4:1) to afford pure 13 (246mg, 88% yield): [* α *]²⁵p +1.95° (c 1.44, CHCls); IR* **(liquid** *film): 2980, 2940, 1605. 1470, 1370, 1195, 1180, 1140 'H-NMR-(CDCl3)-6 0.9 (3 H, d, i=S.S He, CH3CH), 1.2 (3 H, t, J=7 Hi, cm-l;* CH_3CH_2O), 1.3 (3 H, d, <u>J</u>=5.5 Hz, OCH(C<u>H</u>_3)O, 1.6 (3 H, broad s, CH_3C=),
1.0-2.2 (7 H, m, 3CH_2 and CH), 2.45 (3 H, s, CH_3Ar), 3.5 (2 H, 2q, <u>J</u>=7 Hz,
OC<u>H</u>_2CH_3), 3.85(2 H, 2d, <u>J</u>=5.5 Hz, CH_2OTs), 4.05 (2 H, 2d, *=CHC~rO), 4.7 (1 H, q, J=5.5 Hz, OC~(CHs)O), 5.3 (1 H, tq, J=7 Hz, ss=1.5 Hz, =C&CHz0), 7.3 (2 H, d, J=S Hz, 2 Ar* **orthp to CH3), 7.8 (2** *H, d, J=S Hz, 2 Ar prtho to SOsR); MS m/e* (relative *intensity): 398 (H',l), 341 (7), 325 (40), 309 (ii), 213 191, 185 (121, 155 168), 91 (100)) 89 (26),* **73 (91), 57 (80). Anal.** *Calcd for Cz1H34SOs: C, 63.28; H, 8.60. Found: C, 62.60; H, 8.58.*

 $\frac{R}{d} = \frac{1}{2} \sum_{p=1}^{n} \frac{1}{2} \$ (100). Anal. Calcd for C10H22O: C, 75.88; H, 14.01. Found: C, 75.62; H, 13.88.

(R)-(-)-1-Bromo-3.7-dimethyloctane (I). I was prepared as in ref.15 (85%

yield): [a]*5, -5.5° (c 1.4, CHCl3) [11t. -5.7° (neat)]¹⁵; IR (11

magnesium currents varies solution of the tosylate 12 (167.2mg, 0.42mmol) in
dry THF (1.2mL), cooled at -35°C and under nitrogen, the Grignard solution (1mL,
0.78M) was added dropwise, followed by a 0.1M solution of Li2Cu and finally at room temperature for 30min. The reaction mixture was treated with a
saturated solution of NH4Cl (10mL) and the product isolated by extraction with ether in the usual manner. Flash chromatography of the crude product (hexane:ethyl acetate 3:1) afforded pure 13 (125mg, 72% yield): $[a]^{25}p - 0.45^*$, $[a]^{25}s + 0.52^*$, $[a]^{25}s + s - 0.9^*$ (c=2.3, CHCl3); ¹H-NMR (CDCl3) δ 0.87 (12 H, d, -0.52', [a]²⁵436 -0.9' (c=2.3, CHCl3); ¹H-NMR (CDCl3) δ 0.87 (12 H, d,

J=7 Hz, 4CH3CH), 1.2 (3 H, t, J=7 Hz, CH3CH2O), 1.3 (3 H, d, J=5.5 Hz,

OCH(CH3)0), 1.6 (3 H, s, CH3C=), 1.0-1.8 (13 H, m, 8CH2 and 3CH), 2.0

 $(E) - (7R, 11R) - 3.7.11.15 - Tetramethyl - 2-hexadecen-1-ol (Phxtol) (1).$ solution of HCl 0.5N (2.5mL) was slowly added to a solution of 13 (97mg,
0.263mmol) in THF (2.5mL) at 0°C. The reaction mixture was then stirred for 3h at
room temperature. After addition of a solution of NaHCO3 (5%, 2mL) work up gave crude 1 which was flash chromatographed (hexane:ethyl acetate 1:1) to
give pure phytol (67mg, 86% yield). Parallel deprotection experiments on protected geraniol and nerol afforded pure geraniol and nerol respectively without double bond isomerisation.

bond isomerisation.

1. (1iquid film): 3300, 1665 cm⁻¹; ¹H-NMR (CDCl3) δ 0.9 (12 H, d, J=7 Hz,

1. (2iquid film): 3300, 1665 cm⁻¹; ¹H-NMR (CDCl3) δ 0.9 (12 H, d, J=7 Hz,

4CH₃CH), 2.0 (2 H, t, J=7 Hz, CH₂

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